A New Route for Preparation of 5-Deoxy-5-(hydroxyphosphinyl)- D-mannopyranose and -L-gulopyranose Derivatives

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Dedicated to Professor Wolfgang Pfleiderer on the occasion of his 75th birthday

Starting from methyl 2,3-O-isopropylidene-a-D-mannofuranoside (5), methyl 6-O-benzyl-2,3-O-isopropylidene- α -D-lyxo-hexofuranosid-5-ulose (12) was prepared in three steps. The addition reaction of dimethyl phosphonate to 12, followed by deoxygenation of 5-OH group, provided the 5-deoxy-5-dimethoxyphosphinyl- α -D-mannofuranoside derivative 15a and the β -L-gulofuranoside isomer 15b. Reduction of 15a and 15b with sodium dihydrobis(2-methoxyethoxy)aluminate, followed by the action of HCl and then H_2O_2 , afforded the D mannopyranose (17) and *L*-gulopyranose analog 21, each having a phosphinyl group in the hemiacetal ring. These were converted to the corresponding 1,2,3,4,6-penta-O-acetyl-5-methoxyphosphinyl derivatives 19 and **23**, respectively, structures and conformations $({}^{4}C_{1}$ or ${}^{1}C_{4}$, resp.) of which were established by ${}^{1}H\text{-NMR}$ spectroscopy.

Introduction. - We have prepared various sugar analogs having a P-atom in the hemiacetal ring (phospha sugars) [1] because of considerable interest in the physicochemical properties and potential biological activity, as in the case of aza sugars [2] and thia sugars [3]. Thus, a large number of phospha sugars were synthesized, such as those of D-glucose $(1a,b)$ [4] [5], D-mannose (2) [6], D-galactose (3) [7], and --fucose (4) [8].

For example, the first synthesis of 5-deoxy-5-(hydroxyphosphinyl)-p-mannopyranose (2) was performed starting from methyl 2,3-O-isopropylidene- α -D-mannofuranoside (5) by the sequence of $5 \rightarrow 6 \rightarrow 7 \rightarrow 8 \rightarrow 2$ in ten steps (*Scheme 1*) [6]. Although the introduction of a phosphinyl group at $C(5)$ was accomplished by the addition of dimethyl phosphonate to the nitro olefin 6 with relatively good diastereoselectivity $(86:14)$, the conversion of the 6-NO₂ group of the major isomer 7 to a 6-OH group

resulted in a low yield of 8 because of the simultaneous production of various byproducts.

We have recently found an alternative new procedure to introduce a phosphinyl group into a sugar skeleton: namely, addition of a phosphonate to hexofuranos-5-ulose derivatives and the subsequent deoxygenation of 5-OH group [5] [7]. As the use of such procedures was proved to be effective for preparation of D-glucopyranose and Dgalactopyranose analogs, 1b and 3, respectively, we have decided to employ the new method for preparation of D-mannopyranose analogs, 2, as a series for systematic investigation of the stereoselectivity and synthetic efficiency for dehydroxylation of various α -hydroxyphosphonates (5-hydroxy-5-phosphinylhexofuranoses).

Results and Discussion. - Methyl 2,3-O-isopropylidene- α -D-mannofuranoside (5) served as the starting material for preparation of an important key intermediate 12 for the introduction of a phosphinyl group at $C(5)$ (*Scheme 2*). The epoxidation of 5 under *Mitsunobu*'s conditions afforded the 5,6-O-anhydro derivative $9¹$ (89%) which was treated with BnOH and NaH in 1,2-dimethoxyethane (DME) to give the 6-O-benzyl compound 112) in 94% yield. As an alternative way for preparation of 11, compound 5 was converted to the 6-O-Ts derivative 10 [8] in 96% yield. The treatment of 10 with BnOH and NaH in DME afforded 11 by a one-pot procedure, without isolation of the intermediate 9, in 93% yield.

The addition reaction of dimethyl phosphonate to 12 in the presence of DBU afforded the $(5R)$ -5-(dimethylphosphinyl)-D-lyxo-hexofuranoside derivative 13a (76%) and its $(5S)$ -epimer 13b (19%) . The major $(5R)$ -epimer 13a was converted to the methoxalyl esters 14a with methoxalyl chloride in the presence of 4-(dimethylamino)pyridine (DMAP) and then reduced with $Bu₃SnH$ in the presence of AIBN [11], affording an 81 : 19 mixture of the 5-deoxy products. On structural assignment of the resulting two separable diastereoisomers by ¹H-NMR, it turned out that the major isomer was not the expected 5-deoxy-5-(dimethylphosphinyl)- α -D-mannofuranoside derivative **15a** (16% from **13a**) but the β -L-gulofuranoside isomer **15b** (70%).

The α -D-manno configuration for **15a** was assigned on the basis of the large $J(4.5)$ value (10.7 Hz) and the presence of long-range coupling, $5J(1,\text{P})$ (1.5 Hz) [8][12] (*Fig. 1*)

¹) Compound **9** had been obtained from **5** *via* the 6-O-naphthalenesulfonyl derivative in 45% overall yield [9].

²) Compound 11 had been obtained as a minor product (35% yield) from 5 in two-phase BnBr/aq. NaOH system [10].

and Table 1). Similarly, the β -L-gulo configuration for **15b** was derived from the large $J(4,5)$ value (10.4 Hz), and the presence of $5J(2,P)$ (1.2 Hz) and $4J(3,P)$ (1.1 Hz). Although 5-OH compounds 13a and 13b have no H-atom at $C(5)$, their configurations at $C(5)$ were assigned by comparison to the corresponding 5-deoxy compounds 15a and 15b, respectively, because the similar characteristic tendency of the corresponding coupling constants and the chemical shifts is expected due to almost identical conformations³). Thus, $(5R)$ -configuration for 13a and $(5S)$ -configuration for 13b were derived from the presence of $5J(1,P)$ (for **13a**), and $5J(2,P)$ and $4J(3,P)$ (for **13b**).

Fig. 1. The most favorable conformations for 13a,b and 15a,b

³) The antiperiplanar orientation of $H-C(4)$ and $H-C(5)$ in 15a,b is due to steric interaction around $C(4) - C(5)$ bond, whereas the same orientation of $H-C(4)$ and $HO-C(5)$ in 13a,b could be explained in terms of the intramolecular H-bond between the OH group, and O(3) and/or O(4) [13].

	δ /ppm										
	$H - C(1)$	$H-C(2)$	$H - C(3)$	$H - C(4)$	$H - C(5)$						
13a	4.91	4.57	5.00	4.32	—						
13 _b	4.97	4.49	4.85	4.32	—						
15a	4.81	4.54	4.77	4.29	2.71						
15 _b	4.88	4.48	4.59	4.29	2.62						
	J/Hz										
	J(1,2)	5J(1,P)	J(2,3)	5J(2,P)	J(3,4)	$^{4}J(3,\mathbf{P})$	$J(4,\mathbf{P})$	J(4,5)			
13a	$\overline{0}$	2.1	5.8	$\overline{0}$	2.8	$\mathbf{0}$	1.0				
13 _b	Ω	$\overline{0}$	5.8	1.0	3.4	0.9	4.2				
15a	$\mathbf{0}$	1.5	5.5	$\overline{0}$	3.1	$\mathbf{0}$	5.8	10.7			
15 _b	$\mathbf{0}$	$\boldsymbol{0}$	5.5	1.2	3.1	1.1	7.3	10.4			

Table 1. Selected ¹H-NMR Parameters for Compounds **13a,b** and **15a,b** in CDCl₃

Similarly, the minor (5S)-epimer 13b was converted to 14b, which afforded 15a,b in almost the same ratio and yields as those from 13a. These results, therefore, indicated that an epimerization took place at $C(5)$ *via* a radical intermediate during the reduction of the methoxalyl esters 14a,b [5][7].

As for the predominant production of the L -gulofuranoside $(15b)$ by the radical reduction of 14a,b, we propose transition state A of the radical intermediate from the viewpoint of electronic factors $(Fig. 2)$. Namely, the opposition of the phosphinyl group and electronegative O-atom in the furanose ring reduces intramolecular electrostatic repulsion [14]. Moreover, the alignment of the $\sigma(C(4)-C(3))$ bond with the radical p orbital stabilizes the transition state [15]. Although the mechanistic proposals have been reported for the radical-mediated reduction of α -bromo- β -alkoxycarboxylates [16], no report seems to exist, to the best of our knowledge, for the corresponding β alkoxyphosphonate derivatives. Systematic mechanistic studies concerning stereoselectivity of the reduction for 5-phosphinylhexofuranoses are in progress.

The minor α -D-mannofuranoside 15a was then reduced with sodium dihydrobis(2methoxyethoxy)aluminate (SDMA) to give the 5-phosphino derivative 16, which, with HCl in aq. i-PrOH followed by oxidation with H_2O_2 , afforded 6-O-benzyl-5-deoxy-5-(hydroxyphosphinyl)- α/β -D-mannopyranoses 17 (*Scheme 3*). For the purpose of purification and characterization, compounds 17 were converted to the corresponding

Fig. 2. A plausible conformation for the radical intermediate A and the direction of reduction

5-(methoxyphosphinyl) 1,2,3,4-tetra-O-acetates 18 by treatment with Ac_2O /pyridine and then ethereal CH_2N_2 . As the separation of a diastereoisomeric mixture of 18 was still difficult, unambiguous structural assignment was made by further conversion of 18 to the 1,2,3,4,6-penta-O-acetyl derivatives 19^4). Namely, debenzylation⁵) of 18 by the catalytic hydrogenation over 20% $Pd(OH)/C$, followed by acetylation, afforded the pentaacetates 19. After chromatographic purification, 1,2,3,4,6-penta-O-acetyl-5 deoxy-5-[(R)-methoxyphosphinyl]- α -D-mannopyranose (19a; 6.2% from 15a), its β anomer 19b (8.3%), the 5-[(S)-methoxyphosphinyl]- α -isomer 19c (4.7%), and its β isomer 19d (5.9%) were obtained⁶).

The similar treatment of the major β -L-gulofuranoside **15b** afforded 6-O-benzyl-5deoxy-5-(hydroxyphosphinyl)- α/β -L-gulopyranoses (21) via the 5-phosphino compound 20. The L-gulopyranose analogs 21 were also converted to 5-(methoxyphos-

⁴⁾ Penta-O-acetates are apparently more valuable in view of synthesizing unsubstituted phospha sugars, because it is easy to convert them to deacetylated compounds with MeONa/MeOH.

⁵⁾ On debenzylation of crude 17 by catalytic hydrogenation, a considerable amount of starting material remained unchanged despite many trials. However, the same reaction of 18, which had been purified by column chromatography, proceeded with quantitative yield.

⁶⁾ Compounds 19c/d, 23a/d, and 23b/c were obtained as inseparable mixtures. The yield of each product was based on the ¹ H-NMR.

phinyl) pentaacetates 23 via 22: 1,2,3,4,6-penta-O-acetyl-5-deoxy-5- $[(R)$ -methoxyphosphinyl]- β -L-gulopyranose (23a; 12% from 15b), its α -anomer 23b (5.1%), the 5-[(S)-methoxyphosphinyl]- β -isomer 23c (11%), and its α -anomer 23d (6.5%)⁶).

The precise parameters were obtained for these eight isomers, $19a - d$, $23a - d$ by the analysis of their 500-MHz ¹H-NMR spectra (*Table 2*). Some characteristic features of new products **19b** and $23a - d$ are discussed here in detail for comparison with those of the previously reported isomers **19a,c**, and **d** [6].

	δ /ppm								
	$H - C(1)$	$H - C(2)$	$H - C(3)$	$H - C(4)$	$H - C(5)$	$H - C(6)$	$H' - C(6)$	$MeO-P$	
19a	5.45	5.36	5.34	5.63	2.65	4.49	4.41	3.75	
19b	5.32	5.61	5.14	5.62	2.47	4.50	4.47	3.82	
19c	5.35	5.42	5.25	5.52	2.74	4.58	4.31	3.85	
19d	5.42	5.67	5.11	5.56	2.49	4.58	4.33	3.95	
23a	5.78	5.40	5.44	5.30	2.83	4.40	4.35	3.94	
23 _b	5.63	5.45	5.37	5.46	2.99	4.41	4.38	3.86	
23c	5.49	5.57	5.53	5.36	2.71	4.40	4.40	3.79	
23d	5.61	5.67	5.52	5.49	2.81	4.43	4.39	3.87	
	J/Hz								
	J(1,2)	$J(1,\mathbf{P})$	J(2,3)	$J(2,\mathbf{P})$	J(3,4)	J(4,5)	$J(4,\mathbf{P})$	$J(5,\mathbf{P})$	others
19a	6.4	8.8	2.8	21.6	8.6	9.9	8.3	15.4	
19b	3.1	8.6	2.5	24.4	8.9	10.8	6.8	13.3	
19c	5.5	10.1	3.0	25.1	9.3	10.8	4.8	14.6	
19d	3.6	5.9	2.9	28.9	9.9	11.1	3.9	13.3	
23a	11.3	3.4	2.7	3.1	4.6	3.7	35.4	15.1	a)
23 _b	3.4	14.4	3.0	7.3	6.0	4.3	26.0	18.9	b_1
23c	11.0	4.9	2.6	4.9	5.5	3.7	33.6	15.0	\mathfrak{c}_1
23d	3.4	9.8	3.1	20.1	8.6	4.2	11.0	22.1	
		^a) ${}^4J(3,\mathbf{P}) = 1.8$ Hz. ^b) ${}^4J(1,3) = 1.3$ Hz. ^c) ${}^4J(3,\mathbf{P}) = 1.9$ Hz.							

Table 2. Selected ¹H-NMR Parameters for Compounds $19a-d$ and $23a-d$ in CDCl₃

The large $J(4,5)$ values $(10-11 \text{ Hz})$ of **19a**-d and the small $J(4,5)$ values (ca. 4 Hz) of $23a-d$ indicate the D-mannopyranose configuration of the former and the Lgulopyranose configuration of the latter. Conformations of these compounds (in CDCl₃ solution) are derived from the magnitudes of $J(2,P)$ (20–29 Hz for 19a–d and 23d vs. $3 - 7$ Hz for $23a - c$) and $J(4, P)$ (4-11 Hz for 19a-d and 23d vs. $26 - 35$ Hz for $23a - c$) with respect to the corresponding vicinal dihedral angles. Thus, $19a - d$ and $23d$ are assigned to be predominantly in the ⁴C₁ conformation, whereas **23a** – c exist in the ¹C₄ conformation. Compounds 19a and 23d have smaller $J(2,P)$ (21.6 and 19.8 Hz) and larger $J(4, P)$ values (8.3 and 11.0 Hz) than those of **19b** – **d**, indicating that they exist as a conformational mixture of 4C_1 and 1C_4 forms (*Fig. 3*). By employing the additivity rule for vicinal coupling constants [17], the equilibrium population of 4C_1 and 1C_4 conformers were estimated to be $76:24$ for $19a$ and $81:19$ for $23d^7$).

⁷) The ratios of the conformers were estimated by means of the following values: **19a**: $J(1,2) = 4.0$, $J(2,3) =$ 3.1, $J(3,4) = 9.8$, $J(4,5) = 11.55$ for the pure ⁴C₁ form; $J(1,2) = 11.3$, $J(2,3) = 3.1$, $J(3,4) = 3.6$, $J(4,5) = 3.5$ for the pure ¹C₄ form. **23d**: $J(1,2) = 3.5$, $J(2,3) = 3.1$, $J(3,4) = 9.8$, $J(4,5) = 4.5$ for the pure ⁴C₁ form; $J(1,2) = 2.7$, $J(2,3) = 3.1, J(3,4) = 3.6, J(4,5) = 3.0$ for the pure ¹C₄ form [17].

Fig. 3. Conformational equiliblia for 19a and 23d

As for the D-mannopyranose derivative, the $H-C(3)$ and $H-C(5)$ signals of 19a,c appear downfield from those of 19b,d, thus indicating that the orientation of $AcO-C(1)$ group of 19a,c is axial and that of 19b,d equatorial. For the L-gulopyranose derivatives, the large magnitudes of $J(1,2)$ indicate axial $H-C(1)$ orientation for 23a.c. whereas the smaller $J(1,2)$ value points out the equatorial $H-C(1)$ orientation for 23b.

A slight downfield shift of H-C(4) signal of 19a,b compared with those of 19c,d indicates the axial P=O orientation for **19a,b** and the equatorial P=O orientation for **19c,d.** Likewise, a similar downfield shift of $H-C(2)$ signal of 23c compared with those of 23a,b indicates the axial P=O orientation for 23c and the equatorial P=O orientation for 23a,b. An appreciable downfield shift observed for $H-C(2)$ of 23d could be explained by its equatorial orientation (contrary to the axial $H-C(2)$ for $23a - c$).

The conformation of 23d in favor of 4C_1 is most likely ascribed to the presence of strong destabilizing interactions between 1,3-syn-diaxial $P=O$ and $AcO-C(4)$ group in the ${}^{1}C_{4}$ form as well as those between AcO-C(1) and AcO-C(3) group. The electronic destabilizing effects of axial $P=O$ group for 23d seems to be much stronger than those of the corresponding $P-OMe$ group for 23b, because the latter exists in the 1C_4 form.

Such a difference in the destabilization effect between $P=O$ and $P-OMe$ group is also observed between 19a and 19c. Namely, the repulsion between the axial $P-OME$ and AcO $-C(2)$ groups of **19c** (in the ⁴ C_1 form) scarcely affects the inversion of the chair conformer, while the repulsion between axial $P=O$ and $AcO-C(2)$ group of 19a (in the ⁴ C_1 form) causes interconversion into the ¹ C_4 form at a considerable rate (but still in favor of the 4C_1 form).

For β -D-mannopyranose 19b and β -L-gulopyranose 23c, both having the equatorial $A_cO-C(1)$ group, notable conformational equiliblia are not observed, although there exist destabilizing interactions between the $1,3$ -syn-diaxial $P=O$ and AcO groups.

These precise ¹H-NMR parameters of eight diastereoisomers obtained by the present study are thought to be of high value in determining configurations and conformations for other hexopyranose phospha sugars. Although the desired Dmannofuranose precursor was not obtained as a major product for introduction of phosphinyl group, total reaction steps of the present work are fewer, and yields of the ring-transposition products were higher than by the method previously employed. Extension of this work, including investigation of stereoselectivity of the dehydroxylation for α -hydroxyphosphonates, having other hexopyranose structures, as well as biological evaluation of ɒ-manno- and L-gulopyranose phospha sugars, is anticipated to be of interest.

Experimental Part

General. All reactions were monitored by TLC (Merck silica gel 60F, 0.25 mm) with an appropriate solvent system. Column chromatography (CC) was performed with Katayama silica gel 60K070. Components were detected by exposing the plates to UV light and/or spraying them with 20% H₂SO₄/EtOH (with subsequent heating). The NMR spectra were measured in CDCl₃ with Varian VXR-500 (500 MHz for ¹H) and VXR-200 (81 MHz for ³¹P) spectrometers at 22°; chemical shifts are reported as δ values [ppm] relative to TMS (internal standard for ¹H) and 85% phosphoric acid (external standard for ³¹P). The MS spectra were recorded on a VG-70SE instrument and are given in terms of m/z (relative intensity) compared with the base peak.

Methyl 5,6-Anhydro-2,3-O-isopropylidene-a-D-mannofuranoside (9) [9]. To a soln. of 5 (1.04 g, 4.44 mmol) and Ph3P (1.20 g, 4.57 mmol) in dry toluene (10 ml) was added diethyl azodicarboxylate (DEAD; 0.750 ml, 4.82 mmol). The mixture was refluxed for 8 h and evaporated in vacuo. The residue was purified by CC with AcOEt/hexane 1:4 \rightarrow 1:2 to give 9 (854 mg, 89%). Colorless syrup. R_f (AcOEt/hexane 1:2) 0.48. ¹H-NMR $(500 \text{ MHz}): 1.32, 1.47 \text{ } (2s, \text{ Me}_2\text{C}); 2.77 \text{ } (dd, J(6,6') = 5.2, J(5,6') = 2.8, \text{ H}' - \text{C}(6)); 2.90 \text{ } (dd, J(5,6) = 4.0,$ $H-C(6)$; 3.29 (ddd, J(4,5) = 6.1, H-C(5)); 3.30 (s, MeO-C(C(1)); 3.64 (dd, J(3,4) = 3.7, H-C(4)); 4.57 $(d, J(2,3) = 5.8, J(1,2) = 0, H-C(2))$; 4.81 $(dd, H-C(3))$; 4.90 $(s, H-C(1))$.

Methyl 6-O-Benzyl-2,3-O-isopropylidene- a -D-mannofuranoside (11) [10]. a) From 9. To a suspension of NaH (60% in mineral oil, 250 mg, 6.25 mmol) and BnOH (1.30 ml, 1.26 mmol) in DME (2.0 ml), a soln. of 9 (270 mg, 1.25 mmol) in DME (1.0 ml) at 0° was added. The mixture was stirred at 60° for 5 h, diluted with sat. $NH₄Cl$ (20 ml), and extracted with CHCl₃ three times. The combined org. layers were washed with H₂O, dried (Na₂SO₄), and concentrated in vacuo. The residue was purified by CC with AcOEt/hexane 1:1 to give 11 (380 mg, 94%). Colorless syrup. R_f (AcOEt/hexane 1:2) 0.33. ¹H-NMR (500 MHz): 1.32, 1.47 (2s, Me₂C); 2.75 (br. s, HO-C(5)); 3.28 (s, MeO-C(1)); 3.65 (dd, $J(6,6') = 9.8$, $J(5,6) = 5.8$, H'-C(6)); 3.76 (dd, $J(5,6) = 3.7$, $H-C(6)$; 3.96 (dd, $J(4,5) = 8.2$, $J(3,4) = 3.7$, $H-C(4)$); 4.14 (ddd, $H-C(5)$); 4.55 (d, $J(2,3) = 5.8$, $J(1,2) = 0$, $H-C(2)$); 4.57, 4.62 (2d, ²J = 11.9, 1 H each, CH₂O – C(6)); 4.83 (dd, H – C(3)); 4.88 (s, H – C(1)); 7.30 (m, H_p of Ph); 7.35 $(m, 2 H_o, 2 H_m$ of Ph).

b) From 10 [18]. To a soln. of 10 (4.01 g, 10.4 mmol) dissolved in DME (40 ml), NaH (60% in mineral oil, 450 mg, 11.2 mmol) at 0° was added. After stirring for 10 min, BnOH (8.60 ml, 83.1 mol) and then NaH (1.70 g, 42.5 mmol) were added. The mixture was stirred at 60° for 4 h and worked up by employing the same procedures described above to give 11 (3.12 g, 93%).

Methyl 6-O-Benzyl-2,3-O-isopropylidene-a-D-lyxo-hexofuranosid-5-ulose (12). To a soln. of oxalyl chloride $(2.00 \text{ ml}, 22.9 \text{ mmol})$ in CH₂Cl₂ (20 ml), a soln. of DMSO (3.30 ml, 46.5 mmol) in CH₂Cl₂ (10 ml) at -60° was added under Ar, and then a soln. of 11 (2.91 g, 8.98 mmol) in CH₂Cl₂ (10 ml) was added. The mixture was stirred at -60° for 8 h, and then Et₃N (TEA; 8.0 ml, 57 mmol) was added, followed by stirring at $0-10^{\circ}$ for 0.5 h. The mixture was diluted with CHCl₃ (30 ml) and washed with aq. NaCl soln. The aq. layer was extracted twice with CHCl₃. The combined org. layers were washed once with H₂O, dried (Na₂SO₄), and evaporated in vacuo. The residue was purified by CC to give 12 (2.72 g, 94%). Colorless prisms. M.p. 66–67 $^{\circ}$ (AcOEt/hexane 1:2). R_f $(ACOEt/hexane 1:2)$ 0.52. ¹H-NMR (500 MHz): 1.27, 1.34 (2s, Me₂C); 3.33 (s, MeO-C(1)); 4.34 $(s, CH_2O-C(6))$; 4.54 $(d, J(2,3) = 5.8, J(1,2) = 0, H-C(2))$; 4.61 $(d, {}^{2}J = 11.9, H'-C(6))$; 4.64 $(d, H-C(6))$; 4.66 $(d, J(3,4) = 4.3, H-C(4))$; 5.01 $(s, H-C(1))$; 5.07 $(dd, H-C(3))$; 7.32 $(m, H_p$ of Ph); 7.34 - 7.38 $(m, 2H_q$ and 2 H_m of Ph). FAB-MS: 323 (7.8, $[M+1]^+$), 307 (5.6), 291 (11), 233 (22), 201 (35), 181 (56), 91 (100). HR-MS: $323.1480\ ([M+1], C_{17}H_{23}O_6^*$; calc. 323.1495). Anal. calc. for $C_{17}H_{22}O_6$ (322.36): C 63.34, H 6.88; found: C 63.22, H 6.95.

Methyl (5R)- and (5S)-6-O-Benzyl-5-(dimethoxyphosphinyl)-2,3-O-isopropylidene-a-D-lyxo-hexofuranosides (13a,b). DBU (0.800 ml, 5.25 mmol) was dropwise added to a soln. of 12 (1.21 g, 3.75 mmol) in dimethyl phosphonate (10.0 ml, 90 mmol) at 0° , and the soln. was stirred at this temp. for 0.5 h under Ar. The mixture was treated with sat. NH₄Cl at 20° for 4 h and extracted with CHCl₃ three times. The combined org. layers were washed with H₂O, dried (Na₂SO₄), and evaporated in vacuo. The residue was separated by CC with AcOEt/ hexane $1:1 \rightarrow 2:1$ to give **13a** and **13b**.

Data of 13a: Colorless needles (1.31 g, 76%). M.p. $85-86^{\circ}$ (AcOEt/hexane 2:1). R_f (AcOEt/hexane 2:1) 0.25. ¹H-NMR (500 MHz): see *Table 1*; additionally, 1.32, 1.48 (2s, Me₂C); 3.26 (s, MeO-C(1)); 3.76, 3.81 $(2d, J(P, Me) = 10.7, P(OME_2),$; 3.82 $(dd, J(6, P) = 12.8, J(6,6) = 8.6, H' - C(6)$; 3.88 $(dd, J(6, P) = 26.6,$ $H-C(6)$); 4.45, 4.63 (2d, ²J = 11.9, 1 H each, CH₂O – C(6)); 4.75 (s, HO – C(5)); 7.24 (m, H_p of Ph); 7.30 (m, 2 H_m of Ph); 7.32 (m, 2 H_o of Ph). ³¹P-NMR (81 MHz): 25.2. FAB-MS: 433 (35, [M + 1]⁺), 401 (11), 201 (18), 91 (100). HR-MS: 433.1641 ($[M+1]^+$, C₁₉H₃₀O₉P⁺; calc. 433.1628). Anal. calc. for C₁₉H₂₉O₉P (432.41): C 52.78, H 6.76; found: C 52.66, H 6.89.

Data of 13b: Colorless syrup (322 mg, 20%). R_f (AcOEt/hexane 1:2) 0.18. ¹H-NMR (500 MHz): see Table 1; additionally, 1.23, 1.46 (2s, Me₂C); 3.36 (s, Me₂C(1)); 3.80, 3.85 (2d, J(P,Me) = 10.7, P(OMe)₂); 3.85 $(dd, J(6', P) = 19.8, J(6, 6') = 9.8, H'-(6)$; 3.96 $(t, J(6, P) = 10.0, H-C(6)$; 4.36 (br. s, HO-C(5)); 4.58, 4.64 $(2d, {}^{2}J=11.9, 1 \text{ H each, } CH_2O-C(6))$; 7.28 $(m, H_p \text{ of } Ph)$; 7.33 $(m, 2 H_m \text{ of } Ph)$; 7.35 $(m, 2 H_o \text{ of } Ph)$. ³¹P-NMR (81 MHz): 24.1. FAB-MS: 433 (24, $[M+1]^+$), 401 (13), 201 (15), 91 (100). HR-MS: 433.1651, $([M+1]^+$ $C_{19}H_{30}O_9P^+$; calc. 433.1628).

Methyl 6-O-Benzyl-5-(dimethoxyphosphinyl)-2,3-O-isopropylidene-α-D-manno- and β-L-gulofuranoside (15a and 15b, resp.). Methoxalyl chloride (1.00 ml, 10.9 mmol) was added to a soln. of 13a (1.19 g, 2.75 mmol) and DMAP (1.34 g, 11.0 mmol) in dry MeCN (10 ml) at 0°. The mixture was stirred at 0° for 0.5 h under Ar, and the most of solvent was distilled off in vacuo. The residue was treated with aq. NH₄Cl and extracted with CHCl₃ three times. The combined org. layer was washed with H₂O, dried (Na₂SO₄), and evaporated in vacuo to give the $(5R)$ -5-O-methoxalyl derivative 14a as a pale yellow syrup: R_f (AcOEt/hexane 2:1) 0.35.

The crude 14a was co-evaporated with dry toluene and dissolved in the same solvent (10 ml). Bu₃SnH $(1.10 \text{ ml}, 4.09 \text{ mmol})$ and AIBN $(80 \text{ mg}, 0.49 \text{ mmol})$ were added under Ar. The mixture was stirred at 80° for 2 h and then concentrated *in vacuo*. The residue was separated by CC with AcOEt/hexane $1:1 \rightarrow 2:1$ to give **15a** and 15b.

Data of 15a: Colorless syrup (187 mg, 16%). R_f (AcOEt/hexane 2:1) 0.33. ¹H-NMR (500 MHz): see Table 1; additionally, 1.31, 1.42 (2s, Me₂C); 2.71 (dddd, $J(5,P) = 19.2$, $J(4,5) = 10.7$, $J(5,6') = 4.9$, $J(5,6) = 2.8$, $H-C(5)$; 3.26 (s, MeO-C(1)); 3.71, 3.73 (2d, J(P,Me) = 10.7, P(OMe)₂); 3.94 (ddd, J(6',P) = 29.3, J(6,6') = 9.2, $H'-C(6)$); 3.98 (ddd, J(6,P) = 15.3, H-C(6)); 4.55, 4.61 (2d, ²J = 11.9, 1 H each, CH₂O-C(6)); 7.26 (*m*, H_p of Ph), 7.32 (m, 2 H_m of Ph), 7.35 (m, 2 H_n of Ph). ³¹P-NMR (81 MHz): 31.3. FAB-MS: 417 (21, [M + 1]⁺), 401 (12) , 385 (11) , 177 (19) , 137 (11) , 91 (100) . HR-MS: 417.1691 $([M+1]^+, C_{19}H_{30}O_8P^+;$ calc. 417.1679).

Data of **15b**: Colorless syrup (803 mg, 70%). R_f (AcOEt/hexane 2:1) 0.26. ¹H-NMR (500 MHz): see Table 1; additionally, 1.25, 1.40 (2s, Me₂C); 2.62 (ddt, $J(5, P) = 18.3$, $J(4,5) = 10.4$, $J(5,6) = 4.0$, $J(5,6) = 3.7$, $H-C(5)$; 3.32 (s, MeO-C(1)); 3.74, 3.76 (2d, J(P,Me) = 10.7 Hz, P(OMe)₂); 3.82 (ddd, J(6,P) = 28.1, J(6,6) = 9.8, H' – C(6)); 3.88 (ddd, J(6,P) = 10.6, H – C(6)); 4.51, 4.57 (2d, ²J = 11.9, 1 H each, CH₂O – C(6)); 7.28 (m, H_p of Ph); 7.33 $(m, 2H_0$ and 2 H_m of Ph). ³¹P-NMR (81 MHz): 32.4. FAB-MS: 417 (32, $[M+1]^+$), 401 (18), 385 (15) , 177 (15), 91 (100). HR-MS: 417.1682 ($[M+1]^+$, C₁₉H₃₀O₈P⁺; calc. 417.1679).

By the same procedures described above, 13b (262 mg) was converted to 15a (38.3 mg, 15%) and 15b $(172 \text{ mg}, 68\%)$ *via* intermediate 14b.

1,2,3,4,6-Tetra-O-acetyl-5-deoxy-5-[(R)- and (S)-methoxyphosphinyl]- α / β -D-mannopyranoses $19a - d$. To a soln. of 15a (196 mg, 0.471 mmol) in dry toluene (2.0 ml) was added, with stirring, a soln. of sodium dihydrobis(2-methoxyethoxy)aluminate (SDMA; 70% in toluene, 2.5 ml, 0.85 mmol) in dry toluene (1.0 ml) in small portions during 30 min at -10° under Ar. The stirring was continued at this temp. for 30 min. Then, $\rm H_{2}O$ (0.2 ml) was added to decompose excess SDMA, and the mixture was centrifuged. The precipitate was extracted with several portions of toluene. The org. layers were combined and evaporated in vacuo to give methyl 6-O $benzyl-5-deoxy-2,3-O-isopropylidene-5-phosphino- a -D-mannofuranoside (16) as a colorless syrup.$

This syrup was immediately treated with i-PrOH/0.5 μ HCl 2:1 (3.0 ml) at 90 $^{\circ}$ for 2 h under Ar. After cooling, the mixture was evaporated in vacuo. The residue was dissolved in i-PrOH (2.0 ml), treated with 35% H_2O_2 (0.8 ml, 9.3 mmol) at 20 $^{\circ}$ for 12 h and then concentrated *in vacuo* to give crude 6-O-benzyl-5-deoxy-5- $(hydroxyphosphinyl) - \alpha/\beta$ -D-mannopyranoses (17) as a colorless syrup. This was dissolved in dry pyridine (2.0 ml) , and Ac₂O $(1.0 \text{ ml}, 11 \text{ mmol})$ was added at 0° . The mixture was stirred at r.t. for 12 h, diluted with a small amount of cold H₂O, and concentrated in vacuo. The residue was dissolved in EtOH and passed through a column of Amberlite IR-120(H^+) (20 ml). The eluent was evaporated in vacuo, and the residue was methylated with ethereal CH₂N₂ in dry CH₂Cl₂ (2.0 ml) at 0°. After evaporation of the solvent, the residue was separated by CC with AcOEt/hexane $2:1 \rightarrow$ AcOEt to give 1,2,3,4-tetra-O-acetyl-6-O-benzyl-5-deoxy-5- $f(R)$ - and (S)methoxyphosphinyl]- α/β -D-mannopyranoses (18; 72 mg) as a colorless syrup containing a small amounts of unidentified products $(R_f (AcOEt) 0.59 - 0.52)$.

Compounds 18 dissolved in EtOH (2.0 ml) was hydrogenated in the presence of Pd(OH)₂/C (25 mg) at 20^o under atmospheric pressure of $H₂$. After 12 h, the catalysts was filtered off, and the filtrate was evaporated in *vacuo*. The residue was dissolved in dry pyridine (1.0 ml), and Ac_2O (0.25 ml) was added. After stirring at 20^o for 12 h, cold H₂O was added. The mixture was evaporated in vacuo, and the residue was separated by CC with AcOEt/hexane $2:1 \rightarrow$ AcOEt into three *Fractions A – C*.

Fraction A (R_f (AcOEt) 0.41) gave the 5-[(R)-methoxyphosphinyl]- α -D-mannopyranose 19a [6] as a colorless syrup (13.2 mg, 6.2% from 15a; [6]: 6.1% from 8). 31P-NMR (81 MHz): 39.3.

Fraction B (R_f 0.36) gave a colorless syrup (22.6 mg), which consisted of 5-[(S)-methoxyphosphinyl]- α isomer 19c (4.7% from 15a; [6]: 2.4% from 8) and its β -isomer 19d (5.9%; [6]: 3.6% from 8), the ratio being estimated by ¹H-NMR. ³¹P-NMR (81 MHz): 38.8 (for **19c**), 37.7 (for **19d**).

Fraction C (R_f 0.23) gave 5-[*(R)*-methoxyphosphinyl]- β -isomer **19b** as a colorless syrup (17.7 mg, 8.3%) from **15a**). ¹H-NMR (500 MHz): see *Table 2*; additionally, 2.03, 2.07, 2.08, 2.15, 2.18 (5s, 5 AcO); 3.82 $(d, J(\text{P},\text{Me}) = 11.0, \text{MeO}-P)$; 4.47 $(ddd, J(6',P) = 13.5, J(6,6') = 11.8, J(5,6') = 6.6, H'-C(6))$; 4.50 $(ddd, J(6,P) = 11.1, J(5,6) = 6.9, H-C(6))$. ³¹P-NMR (81 MHz): 38.0. FAB-MS: 453 (8.2, $[M+1]^+$), 410 (11), 393 (6), 368 (8), 351 (48), 321 (28), 309 (80), 207 (75), 188 (100), 164 (36). HR-MS: 453.1163 ($[M+1]^+$, $C_{17}H_{36}O_{12}P^+$; calc. 453.1162).

1,2,3,4,6-Tetra-O-acetyl-5-deoxy-5-[(R)- and (S)-methoxyphosphinyl]- α/β -L-gulopyranoses (23a-d). The procedures similar to those for the preparation of compounds 19 from substrates 15a were employed. Thus compound 15b (210 mg, 0.504 mmol) was converted to $1,2,3,4$ -tetra-O-acetyl-6-O-benzyl-5-deoxy-5- $f(R)$ - and (S) -methoxylphosphoninyl]- α/β -L-gulopyranoses (22) via intermediates 20 and 21. The diastereoisomeric mixture 22 was debenzylated and then acetylated again to give 23. The crude product 23 was separated by CC into Fractions A and B.

Fraction A (R_f (AcOEt) 0.44) gave a colorless syrup (41.1 mg), which consisted of 5-[(R) -methoxyphosphinyl]- β -isomer 23a (12% from 15b) and 5-[(S)-methoxyphosphinyl]- α -isomer 23d (6.5%), the ratio being estimated by ¹H-NMR. ¹H-NMR (500 MHz) of 23a: see Table 2; additionally, 1.98, 2.05, 2.14, 2.15, 2.175 (5s, 5 AcO); 3.84 (d, J(P,Me) = 10.7, MeO-P); 4.35 (ddd, J(6,6') = 11.3, J(5,6') = 9.8, J(6',P) = 7.0, H'-C(6)); 4.40 $(dt, J(6,P) = 5.9, J(5,6) = 5.2, H-C(6))$. ¹H-NMR (500 MHz) of **23d**: see *Table 2*; additionally, 2.04, 2.11, 2.12, 2.135, 2.17 (5s, 5 AcO); 3.87 (d, $J(\text{P},\text{Me}) = 10.7$, MeO-P); 4.39 (m⁸), $J(5,6') = 5.2$, H'-C(6)); 4.43 (dt, $J(6,\text{P}) =$ $19.2, J(6,6') = 11.6, J(5,6) = 3.7, H-C(6))$. ³¹P-NMR (81 MHz): 38.4 (for **23a**); 37.7 (for **23d**). FAB-MS: 453 (5.2, $[M+1]^+$, 410 (9.8), 393 (6), 351 (48), 321 (18), 309 (80), 230 (39), 207 (79), 188 (100), 164 (32). HR-MS: 453.1169 ($[M+1]^+$, C₁₇H₃₆O₁₂P⁺; calc. 453.1162).

Fraction B (R_f 0.39) gave a colorless syrup (36.1 mg), which consisted of 5-[(R) -methoxyphosphinyl]- α isomer 23b (5.1% from 15b) and 5-[(S)-methoxyphosphinyl]- β -isomer 23c (11%), the ratio being estimated by ¹H-NMR. ¹H-NMR (500 MHz) for **23b**: see *Table 2*; additionally, 2.06, 2.10, 2.115, 2.12, 2.175 (5s, 5 AcO); 3.86 $(d, J(P, Me) = 11.0, MeO-P)$; 4.40 $(ddd, J(6,6') = 11.6, J(5,6') = 8.5, J(6', P) = 6.7, H'-C(6))$; 4.41 $(ddd, J(6,P) =$ $9.9, J(5,6) = 4.5, H-C(6))$. ¹H-NMR (500 MHz) for **23c**: see *Table 2*; additionally, 2.00, 2.07, 2.16, 2.18, 2.19 (5s, 5 AcO); 3.79 (d, J(P,Me) = 11.0, MeO-P); 4.40 (dd, J(6,P) = J(6',P) = 9.8, J(5,6') = J(5,6) = 7.3, CH₂(6)). ³¹P-NMR (81 MHz): 37.8 (for 23b); 39.1 (for 23c). FAB-MS: 453 (4.2, $[M+1]^+$), 410 (11), 393 (10), 351 (38), 321 (15) , 309 (69) , 230 (32) , 207 (68) , 188 (100) , 164 (39) . HR-MS: 453.1173 $([M+1]^+, C_{17}H_{36}O_{12}P^+$; calc. 453.1162).

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⁸) The $J(6', P)$ value is uncertain because of overlap with other signals.

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